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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/810,630	03/29/2004	Akihiko Shimasaki	KPC-0307	9861
23353	7590	06/21/2006	EXAMINER	
RADER FISHMAN & GRAUER PLLC LION BUILDING 1233 20TH STREET N.W., SUITE 501 WASHINGTON, DC 20036			FEELY, MICHAEL J	
			ART UNIT	PAPER NUMBER
			1712	

DATE MAILED: 06/21/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary	Application No.	Applicant(s)	
	10/810,630	SHIMASAKI ET AL.	
	Examiner Michael J. Feely	Art Unit 1712	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) Responsive to communication(s) filed on 29 March 2004.
- 2a) This action is FINAL. 2b) This action is non-final.
- 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) Claim(s) 1-8 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) Claim(s) _____ is/are allowed.
- 6) Claim(s) 1-8 is/are rejected.
- 7) Claim(s) 3 and 7 is/are objected to.
- 8) Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) The specification is objected to by the Examiner.
- 10) The drawing(s) filed on _____ is/are: a) accepted or b) objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
 - a) All b) Some * c) None of:
 1. Certified copies of the priority documents have been received.
 2. Certified copies of the priority documents have been received in Application No. _____.
 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- 1) Notice of References Cited (PTO-892)
- 2) Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)
Paper No(s)/Mail Date 20040329.
- 4) Interview Summary (PTO-413)
Paper No(s)/Mail Date. _____.
- 5) Notice of Informal Patent Application (PTO-152)
- 6) Other: _____.

DETAILED ACTION

Claim Objections

1. Claims 3 and 7 objected to because of the following informalities: in claim 3, the language "is contained as the curing agent of the coating composition" is redundant. Claim 7 is objected to because it is dependent from claim 3. Appropriate correction is required.

Claim Rejections - 35 USC § 103

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

4. The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

1. Determining the scope and contents of the prior art.
2. Ascertaining the differences between the prior art and the claims at issue.
3. Resolving the level of ordinary skill in the pertinent art.
4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

5. Claims 1-8 are rejected under 35 U.S.C. 103(a) as being obvious over the following references in view of Sawada et al. (US Pat. No. 6,355,351): Nishiguchi et al. (US Pat. No.

6,942,922), Nishiguchi et al. (US Pat. No. 6,734,260), Nishiguchi et al. (US Pat. No. 6,761,973), and Shigeo et al. (US Pat. No. 6,680,122).

6. Claims 1, 2, 4-6, and 8 are rejected under 35 U.S.C. 103(a) as being obvious over Nishiguchi et al. (US Pat. No. 6,660,385) in view of Sawada et al. (US Pat. No. 6,355,351).

The applied references have a common assignee with the instant application; however, the inventive entity is different. Based upon the earlier effective U.S. filing date of the reference, it constitutes prior art only under 35 U.S.C. 102(e). This rejection under 35 U.S.C. 103(a) might be overcome by: (1) a showing under 37 CFR 1.132 that any invention disclosed but not claimed in the reference was derived from the inventor of this application and is thus not an invention “by another”; (2) a showing of a date of invention for the claimed subject matter of the application which corresponds to subject matter disclosed but not claimed in the reference, prior to the effective U.S. filing date of the reference under 37 CFR 1.131; or (3) an oath or declaration under 37 CFR 1.130 stating that the application and reference are currently owned by the same party and that the inventor named in the application is the prior inventor under 35 U.S.C. 104, together with a terminal disclaimer in accordance with 37 CFR 1.321(c). This rejection might also be overcome by showing that the reference is disqualified under 35 U.S.C. 103(c) as prior art in a rejection under 35 U.S.C. 103(a). See MPEP § 706.02(l)(1) and § 706.02(l)(2).

Nishiguchi et al. (US Pat. No. 6,942,922) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1, 16 & 18); (2) a base resin (claims 1, 16 & 18); and (3) a curing agent (claims 1, 16 & 18); wherein the base resin is selected from the group consisting of: (A) a base resin (I), which is a xylene-formaldehyde-resin-modified amino-containing epoxy resin obtained by

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reacting an epoxy resin (1) having an epoxy equivalent of from 180 to 2500 with a xylene formaldehyde resin (2) and an amino-containing compound (3) (claims 1, 16 & 18); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 16, 18 & 22);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1, 16 & 18); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 16, 18 & 22);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (column 6, line 61 through column 7, line 65); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 16, 18 & 22);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1, 16 & 18); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 16, 18 & 22).

Nishiguchi et al. (US Pat. No. 6,734,260) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1, 20 & 22); (2) a base resin (claims 1, 20 & 22); and (3) a curing agent (claims 1, 20 & 22); wherein the base resin is selected from the group consisting of: (A) a base resin (I), which is a xylene-formaldehyde-resin-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent of from 180 to 2500 with a xylene formaldehyde resin (2) and an amino-containing compound (3) (claims 1, 20 & 22); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 20, 22 & 25);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1, 20 & 22); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 20, 22 & 25);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (column 7, line 43 through column 8, line 44); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 20, 22 & 25);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1, 20 & 22); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 20, 22 & 25).

Nishiguchi et al. (US Pat. No. 6,761,973) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1-3 & 8; column 11, line 61 through column 12, line 10); (2) a base resin (claims 1-3 & 8; column 11, line 61 through column 12, line 10); and (3) a curing agent (claims 1-3 & 8; column 11, line 61 through column 12, line 10); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1-3 & 8; column 11, line 61 through column 12, line 10); (5) an article coated with the coating

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composition as claimed in claim 1 (claims 1-3, 8 & 19; column 11, line 61 through column 12, line 10);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1-3 & 8; column 11, line 61 through column 12, line 10); (6) an article coated with the coating composition as claimed in claim 2 (claims 1-3, 8 & 19; column 11, line 61 through column 12, line 10);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (claims 1-3 & 8; column 11, line 61 through column 12, line 10); (7) an article coated with the coating composition as claimed in claim 3 (claims 1-3, 8 & 19; column 11, line 61 through column 12, line 10);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1-3 & 8; column 11, line 61 through column 12, line 10); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1-3, 8 & 19; column 11, line 61 through column 12, line 10).

Shigeo et al. (US Pat. No. 6,680,122) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 11); (2) a base resin (claims 1 & 11); and (3) a curing agent (claims 1 & 11); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a

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caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1 & 11); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 11 & 15);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1 & 11); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 11 & 15);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (column 4, line 38 through column 5, line 16); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 11 & 15);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 11); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 11 & 15).

Nishiguchi et al. (US Pat. No. 6,660,385) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 24-26); (2) a base resin (claims 1 & 24-26); and (3) a curing agent (claims 1 & 24-26); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1 & 24-26); and (C) a base resin (III), which is a polyol-

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modified amino-containing epoxy resin (III) obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with an alkyl phenol (v₁) and/or a carboxylic acid (v₂), a polyol compound (4) available by adding a caprolactone to a compound having a plurality of active hydrogen groups, and a amino-containing compound (3) (claims 1 & 24-26); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 24-26 & 29);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1 & 24-26); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 24-26 & 29);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 24-26); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 24-26 & 29).

With respect to all of these prior art references, they disclose the use of a bismuth compound as a rust preventative. Hence, they do not disclose the use of: (1) a rust preventative (corrosion inhibitor) *selected from corrosion inhibitors of cerium compounds, lanthanum compounds, molybdate salt compounds, gluconic acid derivative salts, porous base materials, triazole compounds, thiazole compounds, tetracyclines, and metal phosphate salt compounds of ascorbic acid.*

Sawada et al. disclose an analogous cationic composition (*see Abstract*), which further includes a rust preventive pigment (*see column 8, lines 10-21*). They disclose, “The pigment described above includes, for example...rust preventive pigments such as basic lead silicate, *aluminum phosphomolybdate*, aluminum tripolyphosphate, strontium chromate, zinc chromate,

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lead chromate, and *bismuth-containing compounds.*" This disclosure demonstrates that aluminum phosphomolybdate (a molybdate salt compound) and bismuth-containing compounds are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. In light of this, it has been found to be *prima facie* obvious to substitute known equivalents used for the same purpose – *see MPEP 2144.06.*

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to substitute the bismuth compound with a molybdate salt compound, as taught by Sawada et al., in the cationic compositions of the above-discussed prior art because the teachings of Sawada et al. demonstrate that these materials are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. The substitution of such equivalents has been found to be *prima facie* obvious.

7. Claims 1-8 are rejected under 35 U.S.C. 103(a) as being unpatentable over the following references in view of Sawada et al. (US Pat. No. 6,355,351): Shigeo et al. (EP 1314768), Nishiguchi et al. (JP 2003-306636), Nishiguchi et al. (2003-221547), Nishiguchi et al. (US Pat. No. 6,503,629), Nishiguchi et al. (US Pat. No. 6,492,027), Shigeo et al. (Pub. No.: US 2002/0119318), Shigeo et al. (EP 1111013), Nishiguchi et al. (JP 2002-060454), and Nishiguchi et al. (JP 2001-279168).

8. Claims 1, 2, 4-6, and 8 are rejected under 35 U.S.C. 103(a) as being obvious over Nishiguchi et al. (Pub. No.: US 2002/0068172) in view of Sawada et al. (US Pat. No. 6,355,351).

Shigeo et al. (EP 1314768) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract;

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paragraphs 0042, 0045 & 0049); (2) a base resin (Abstract; paragraphs 0042, 0045 & 0049); and (3) a curing agent (Abstract; paragraphs 0042, 0045 & 0049); wherein the base resin is selected from the group consisting of: (A) a base resin (I), which is a xylene-formaldehyde-resin-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent of from 180 to 2500 with a xylene formaldehyde resin (2) and an amino-containing compound (3) (Abstract; paragraphs 0042, 0045 & 0049); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0042, 0045 & 0049);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (Abstract; paragraphs 0042, 0045 & 0049); (6) an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0042, 0045 & 0049);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0042, 0045 & 0049); (7) an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0042, 0045 & 0049);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0042, 0045 & 0049); and (8) an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0042, 0045 & 0049).

Nishiguchi et al. (JP 2003-306636) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract; paragraphs 0042-0043 & 0047); (2) a base resin (Abstract; paragraphs 0042-0043 & 0047); and (3) a curing agent (Abstract; paragraphs 0042-0043 & 0047); wherein the base resin

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is selected from the group consisting of: (A) a base resin (I), which is a xylene-formaldehyde-resin-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent of from 180 to 2500 with a xylene formaldehyde resin (2) and an amino-containing compound (3) (Abstract; paragraphs 0042-0043 & 0047); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0042-0043 & 0047);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (Abstract; paragraphs 0042-0043 & 0047); (6) an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0042-0043 & 0047);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0042-0043 & 0047); (7) an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0042-0043 & 0047);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0042-0043 & 0047); and (8) an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0042-0043 & 0047).

Nishiguchi et al. (2003-221547) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract; paragraphs 0044 & 0051); (2) a base resin (Abstract; paragraphs 0044 & 0051); and (3) a curing agent (Abstract; paragraphs 0044 & 0051); wherein the base resin is selected from the group consisting of: (A) a base resin (I), which is a xylene-formaldehyde-resin-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy

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equivalent of from 180 to 2500 with a xylene formaldehyde resin (2) and an amino-containing compound (3) (Abstract; paragraphs 0044 & 0051); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0044 & 0051);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (Abstract; paragraphs 0044 & 0051); (6) an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0044 & 0051);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0044 & 0051); (7) an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0044 & 0051);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0044 & 0051); and (8) an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0044 & 0051).

Nishiguchi et al. (US Pat. No. 6,503,629) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 13-15); (2) a base resin (claims 1 & 13-15); and (3) a curing agent (claims 1 & 13-15); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an

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amino-containing compound (3) (claims 1 & 13-15); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 13-15 & 18);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1 & 13-15); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 13-15 & 18);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (column 6, lines 16-30); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 13-15 & 18);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 13-15); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 13-15 & 18).

Nishiguchi et al. (US Pat. No. 6,492,027) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 20); (2) a base resin (claims 1 & 20); and (3) a curing agent (claims 1 & 20); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1 & 20); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 20 & 23);

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(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1 & 20); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 20 & 23);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (column 7, lines 25-40); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 20 & 23);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 20); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 20 & 23).

Shigeo et al. (Pub. No.: US 2002/0119318) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 11-13); (2) a base resin (claims 1 & 11-13); and (3) a curing agent (claims 1 & 11-13); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1 & 11-13); (5) an article coated with the coating composition as claimed in claim 1 (claims 1, 11-13 & 16);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a

blocking agent (claims 1 & 11-13); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 11-13 & 16);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (paragraphs 0040); (7) an article coated with the coating composition as claimed in claim 3 (claims 1, 11-13 & 16);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 11-13); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 11-13 & 16).

Shigeo et al. (EP 1111013) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract; paragraphs 0032 & 0036); (2) a base resin (Abstract; paragraphs 0032 & 0036); and (3) a curing agent (Abstract; paragraphs 0032 & 0036); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (Abstract; paragraphs 0032 & 0036); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0032 & 0036);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a

blocking agent (Abstract; paragraphs 0032 & 0036); (6) an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0032 & 0036);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0032 & 0036); (7) an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0032 & 0036);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0032 & 0036); and (8) an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0032 & 0036).

Nishiguchi et al. (JP 2002-060454) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract; paragraphs 0034 & 0038); (2) a base resin (Abstract; paragraphs 0034 & 0038); and (3) a curing agent (Abstract; paragraphs 0034 & 0038); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (Abstract; paragraphs 0034 & 0038); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0034 & 0038);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a

blocking agent (Abstract; paragraphs 0034 & 0038); (6) an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0034 & 0038);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0034 & 0038); (7) an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0034 & 0038);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0034 & 0038); and (8) an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0034 & 0038).

Nishiguchi et al. (JP 2001-279168) disclose: (1) a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (Abstract; paragraphs 0031 & 0035); (2) a base resin (Abstract; paragraphs 0031 & 0035); and (3) a curing agent (Abstract; paragraphs 0031 & 0035); wherein the base resin is selected from the group consisting of: (B) a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (Abstract; paragraphs 0031 & 0035); (5) an article coated with the coating composition as claimed in claim 1 (Abstract; paragraphs 0031 & 0035);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a

blocking agent (Abstract; paragraphs 0031 & 0035); **(6)** an article coated with the coating composition as claimed in claim 2 (Abstract; paragraphs 0031 & 0035);

(3) wherein the curing agent is a block polyisocyanate curing agent (II) obtained by reacting an active-hydrogen-containing component propylene glycol with an aromatic polyisocyanate (Abstract; paragraphs 0031 & 0035); **(7)** an article coated with the coating composition as claimed in claim 3 (Abstract; paragraphs 0031 & 0035);

(4) wherein the coating composition is a cationic electrodeposition coating (Abstract; paragraphs 0031 & 0035); and **(8)** an article coated with the coating composition as claimed in claim 4 (Abstract; paragraphs 0031 & 0035).

Nishiguchi et al. (Pub. No.: US 2002/0068172) disclose: **(1)** a coating composition used for coating of a steel material and/or aluminum material, comprising: (1) at least one corrosion inhibitor (claims 1 & 18-20); (2) a base resin (claims 1 & 18-20); and (3) a curing agent (claims 1 & 18-20); wherein the base resin is selected from the group consisting of: **(B)** a base resin (II), which is a polyol-modified amino-containing epoxy resin obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with a polyol compound (4) available by adding a caprolactone to a compound containing a plurality of active hydrogen groups and an amino-containing compound (3) (claims 1 & 18-20); and **(C)** a base resin (III), which is a polyol-modified amino-containing epoxy resin (III) obtained by reacting an epoxy resin (1) having an epoxy equivalent weight of from 180 to 2500 with an alkyl phenol (v_1) and/or a carboxylic acid (v_2), a polyol compound (4) available by adding a caprolactone to a compound having a plurality of active hydrogen groups, and a amino-containing compound (3) (claims 1 & 18-20); **(5)** an article coated with the coating composition as claimed in claim 1 (claims 1, 18-20 & 23);

(2) wherein the curing agent is a curing agent (I), which is a blocked polyisocyanate compound obtained by blocking an isocyanate group of a polyisocyanate compound with a blocking agent (claims 1 & 18-20); (6) an article coated with the coating composition as claimed in claim 2 (claims 1, 18-20 & 23);

(4) wherein the coating composition is a cationic electrodeposition coating (claims 1 & 18-20); and (8) an article coated with the coating composition as claimed in claim 4 (claims 1, 18-20 & 23).

With respect to all of these prior art references, they disclose the use of a bismuth compound as a rust preventative. Hence, they do not disclose the use of: (1) a rust preventative (corrosion inhibitor) *selected from corrosion inhibitors of cerium compounds, lanthanum compounds, molybdate salt compounds, gluconic acid derivative salts, porous base materials, triazole compounds, thiazole compounds, tetracyclines, and metal phosphate salt compounds of ascorbic acid.*

Sawada et al. disclose an analogous cationic composition (*see Abstract*), which further includes a rust preventive pigment (*see column 8, lines 10-21*). They disclose, “The pigment described above includes, for example...rust preventive pigments such as basic lead silicate, *aluminum phosphomolybdate*, aluminum tripolyphosphate, strontium chromate, zinc chromate, lead chromate, and *bismuth-containing compounds*.” This disclosure demonstrates that aluminum phosphomolybdate (a molybdate salt compound) and bismuth-containing compounds are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. In light of this, it has been found to be *prima facie* obvious to substitute known equivalents used for the same purpose – *see MPEP 2144.06*.

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Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to substitute the bismuth compound with a molybdate salt compound, as taught by Sawada et al., in the cationic compositions of the above-discussed prior art because the teachings of Sawada et al. demonstrate that these materials are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. The substitution of such equivalents has been found to be *prima facie* obvious.

Double Patenting

9. The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the “right to exclude” granted by a patent and to prevent possible harassment by multiple assignees. A nonstatutory obviousness-type double patenting rejection is appropriate where the conflicting claims are not identical, but at least one examined application claim is not patentably distinct from the reference claim(s) because the examined application claim is either anticipated by, or would have been obvious over, the reference claim(s). See, e.g., *In re Berg*, 140 F.3d 1428, 46 USPQ2d 1226 (Fed. Cir. 1998); *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) or 1.321(d) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent either is shown to be commonly owned with this application, or claims an invention made as a result of activities undertaken within the scope of a joint research agreement.

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

10. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 16-18, and 22 of U.S. Patent No. 6,942,922 in view of Sawada et al. (US Pat. No. 6,355,351).

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11. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 20-22, and 25 of U.S. Patent No. 6,734,260 in view of Sawada et al. (US Pat. No. 6,355,351).

12. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 11, and 15 of U.S. Patent No. 6,680,122 in view of Sawada et al. (US Pat. No. 6,355,351).

13. Claims 1, 2, 4-6, and 8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 24-26, and 29 of U.S. Patent No. 6,660,385 in view of Sawada et al. (US Pat. No. 6,355,351).

14. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 13-15, and 18 of U.S. Patent No. 6,503,629 in view of Sawada et al. (US Pat. No. 6,355,351).

15. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1, 20, and 23 of U.S. Patent No. 6,492,027 in view of Sawada et al. (US Pat. No. 6,355,351).

With respect to all of these prior art references, they disclose the use of a bismuth compound as a rust preventative. Hence, they do not disclose the use of: (1) a rust preventative (corrosion inhibitor) selected from corrosion inhibitors of cerium compounds, lanthanum compounds, molybdate salt compounds, gluconic acid derivative salts, porous base materials, triazole compounds, thiazole compounds, tetracyclines, and metal phosphate salt compounds of ascorbic acid.

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Sawada et al. disclose an analogous cationic composition (*see Abstract*), which further includes a rust preventive pigment (*see column 8, lines 10-21*). They disclose, “The pigment described above includes, for example...rust preventive pigments such as basic lead silicate, *aluminum phosphomolybdate*, aluminum tripolyphosphate, strontium chromate, zinc chromate, lead chromate, and *bismuth-containing compounds*.” This disclosure demonstrates that aluminum phosphomolybdate (a molybdate salt compound) and bismuth-containing compounds are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. In light of this, it has been found to be *prima facie* obvious to substitute known equivalents used for the same purpose – *see MPEP 2144.06*.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to substitute the bismuth compound with a molybdate salt compound, as taught by Sawada et al., in the cationic compositions of the above-discussed prior art because the teachings of Sawada et al. demonstrate that these materials are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. The substitution of such equivalents has been found to be *prima facie* obvious.

Furthermore, with respect to claims 3 and 7, the specifications of these patents disclose these embodiments as exemplary and preferred. In light of this, it has been found that those portions of the specification which provide support for the patent claims may also be examined and considered when addressing the issue of whether a claim in the application defines an obvious variation of an invention claimed in the patent – *In re Vogel*, 422 F.2d 438, 441-42, 164 USPQ 619, 622 (CCPA 1970). The court in *Vogel* recognized “that it is most difficult, if not meaningless, to try to say what is or is not an obvious variation of a claim,” but that one can

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judge whether or not the invention claimed in an application is an obvious variation of an embodiment disclosed in the patent which provides support for the patent claim. According to the court, one must first “determine how much of the patent disclosure pertains to the invention claimed in the patent” because only “[t]his portion of the specification supports the patent claims and may be considered.” The court pointed out that “this use of the disclosure is not in contravention of the cases forbidding its use as prior art, nor is it applying the patent as a reference under **35 U.S.C. 103**, since only the disclosure of the invention claimed in the patent may be examined,” – *see MPEP 804 II B 1.*

Therefore, the limitations of instant claims 3 and 7 would have been obvious because the patent documents disclose these limitations as exemplary and preferred embodiments of the patented invention.

16. Claims 1-8 are rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined teachings of claims 1-3, 8, and 19 of U.S. Patent No. 6,761,973 in view of Sawada et al. (US Pat. No. 6,355,351).

The claims of this patent do not explicitly disclose the use of corrosion inhibitor; however, the specification (*see column 11, line 60 through column 12, line 10*) discloses the use of a bismuth compound as an exemplary embodiment of the invention (*see In re Vogel*).

As set forth above, the teachings of Sawada et al. demonstrate that aluminum phosphomolybdate (a molybdate salt compound) and bismuth-containing compounds are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic

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composition. In light of this, it has been found to be *prima facie* obvious to substitute known equivalents used for the same purpose – *see MPEP 2144.06.*

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to substitute the bismuth compound with a molybdate salt compound, as taught by Sawada et al., in the cationic compositions of the above-discussed prior art because the teachings of Sawada et al. demonstrate that these materials are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. The substitution of such equivalents has been found to be *prima facie* obvious.

17. Claims 1-8 are provisionally rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over the combined limitations of claims 13, 14, 16, 17, and 19 of copending Application No. 10/810,706 (Pub. No. US 2005/0214470) in view of Sawada et al. (US Pat. No. 6,355,351).

As set forth above, the teachings of Sawada et al. demonstrate that aluminum phosphomolybdate (a molybdate salt compound) and bismuth-containing compounds are recognized in the art as equivalent rust preventive pigments for this type of epoxy-based cationic composition. In light of this, it has been found to be *prima facie* obvious to substitute known equivalents used for the same purpose – *see MPEP 2144.06.*

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to substitute the bismuth compound with a molybdate salt compound, as taught by Sawada et al., in the cationic compositions of the above-discussed prior art because the teachings of Sawada et al. demonstrate that these materials are recognized in the art as equivalent rust

preventive pigments for this type of epoxy-based cationic composition. The substitution of such equivalents has been found to be *prima facie* obvious.

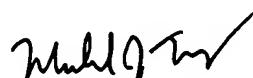
This is a provisional obviousness-type double patenting rejection.

Communication

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Michael J. Feely whose telephone number is 571-272-1086. The examiner can normally be reached on M-F 8:30 to 5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Randy Gulakowski can be reached on 571-272-1302. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.



Michael J. Feely
Primary Examiner
Art Unit 1712

June 11, 2006

**MICHAEL FEELY
PRIMARY EXAMINER**